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<b>(21) International Application Number:</b> PCT/US96/02017 <b>(22) International Filing Date:</b> 13 February 1996 (13.02.96) <b>(30) Priority Data:</b> 08/389,475 13 February 1995 (13.02.95) US <b>(71) Applicant:</b> THE REGENTS OF THE UNIVERSITY OF CALIFORNIA [US/US]; 22nd floor, 300 Lakeside Drive, Oakland, CA 94162-3550 (US). <b>(72) Inventors:</b> CHARYCH, Deborah; 915 Polk Street, Albany, CA 94706 (US). REICHERT, Anke; 704 Stannage, Albany, CA 94706 (US). <b>(74) Agents:</b> ZIMMERMAN, C., Michael et al.; Flehr, Hohbach, Test, Albritton & Herbert, Suite 3400, 4 Embarcadero Center, San Francisco, CA 94111-4187 (US).		<b>(81) Designated States:</b> AL, AM, AT, AU, AZ, BB, BG, BR, BY, CA, CH, CN, CZ, DE, DK, EE, ES, FI, GB, GE, HU, IS, JP, KE, KG, KP, KR, KZ, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, TJ, TM, TR, TT, UA, UG, UZ, VN, ARIPO patent (KE, LS, MW, SD, SZ, UG), Eurasian patent (AZ, BY, KG, KZ, RU, TJ, TM), European patent (AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR, NE, SN, TD, TG).  <b>Published</b> <i>With international search report.</i> <i>With amended claims.</i>
<b>(54) Title:</b> THREE-DIMENSIONAL COLORIMETRIC ASSAY ASSEMBLIES  <b>(57) Abstract</b>  A direct assay is described using novel three-dimensional polymeric assemblies which change from a blue to red color when exposed to an analyte, in one case a flu virus. The assemblies are typically in the form of liposomes which can be maintained in a suspension, and show great intensity in their color changes. Their method of production is also described.		

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## THREE-DIMENSIONAL COLORIMETRIC ASSAY ASSEMBLIES

The present application is a continuation in part of prior-filed U.S. Patent Application Nos. 08/289,384 filed August 11th, 1994, and 08/328,237 filed October 24th, 1994, both of which are continuations in part of 08/159,927 filed November 30th, 1993, which is a continuation in part of both U.S. Patent Application Nos. 07/982,189 filed November 25th, 1992, and 07/976,697 filed November 13th, 1992, both of the later cases being related to U.S. Patent Application No. 07/617,988 filed November 26th, 1990.

This invention was made with Government support under Contract No DE-AC03-76SF00098 between the U.S. Department of Energy and the University of California for the operation of Lawrence Berkeley Laboratory. The Government has certain rights in this invention.

## BACKGROUND OF THE INVENTION

The present invention relates to a method for direct detection of analytes using color changes in three-dimensional polymeric assemblies which occur in response to selective binding of analytes to their surface.

Analytical Chemistry Analytical chemistry techniques have been used for many years to determine such medical parameters as hematocrit levels. While useful in their own right, analytical chemistry methods are of limited or no practical applicability to many biological parameters in which assessment would be valuable. Unless expensive and cumbersome gas chromatography methods are used, large quantities of analytes are generally required to accomplish such methods. Often, quantitative results are limited or not available. However, such techniques have been used for such basic chemical tests as creatinine assays.

Microbiological and Pathology Methods Another approach to medical-biological systems analysis has been direct microscopic observation using various cell-staining and

classic pathology techniques. Augmenting these capabilities have been well developed microbiological techniques, such as culturing, colony characterization, and observation of metabolic and nutrient limitations. Most of medical science  
5 has been developed using this basic arsenal of analytic techniques. While culturing and direct-tissue observation techniques have served as the bulwark of medical detection processes for many years, they have considerable limitations.

10 Pathological analysis of patient tissues to determine the development of a disease state and the identification of the causative pathogen generally requires an invasive procedure. On the other hand, culturing the pathogen from various body fluid or other samples is time consuming and  
15 expensive.

Immunoassays A breakthrough in medicine occurred with the development of immunoassay techniques. In these methods, an antibody is developed which will specifically bind to a target of interest. While costly in both their  
20 development and production, antibodies from animals allowed a very accurate analysis of a number of analytes which had previously been virtually unassessable in both research and particularly clinical situations.

An important technical advancement in immunoassay was  
25 the development of monoclonal antibodies. Instead of subjecting an animal to an analyte and harvesting its whole range of antibodies, in this technique a single spleen cell of a sensitized animal is rendered immortal and multiplied many times. The resulting cell line is then cultured to  
30 produce a very specific and pure antibody product.

Because the antibody itself is a small molecule, it must be labeled in some way so that the binding event can be detected. This can be done with a dye, fluorescent, radioactive or other label. Conversely, if binding

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inhibition occurs between a known amount of introduced, labeled analyte and the material to be analyzed, the diminution of the signal will indicate the presence of test analyte. If the agglutination of the antibody particles is of sufficient volume and density, the formation of a precipitant can also serve to signal the presence of an analyte.

In recent years, the research and medical communities have come to rely heavily on immunoassay techniques to detect and quantify biological materials. While successful in many respects, the indirect nature of immunoassay methods as well as their dependence on antibody materials, results in a variety of complications, problems, and assay limitations. Briefly, the development and production of antibodies remains expensive, and these molecules are sensitive to environmental changes. Also, only those materials to which antibodies can be produced can be detected by these systems.

#### Langmuir-Blodgett Film Assays

The techniques of molecular self-assembly, such as that described by Swalen et al., (Langmuir, Vol. 3, page 932, 1987) as well as Langmuir-Blodgett (LB) deposition (Roberts, Ed. Langmuir-Blodgett Films, Wiley, New York, 1966) have been used for coating surfaces with a well-defined, quasi two-dimensional array of molecules. The initial use for this new advancement was for materials science applications such as wetting (Whitesides, et al., Langmuir, Vol. 6, p. 87, 1990) and friction (Novotny et al., Langmuir, Vol. 5, p. 485, 1989).

These bilayer films are also used as immobilizing supports for analytic reactions. Bio-sensors based on LB films can detect molecules of diagnostic significance such as glucose (Okahata, et al., Thin Solid Films, Vol. 180, p. 65, 1989) and urea (Arisawa, et al., Thin Solid Films, Vol.

210, p. 443, 1992). In these cases, classic analytical chemistry systems are immobilized on the films in order to improve the readout of the test results and otherwise simplify and improve the detection capabilities of the test  
5 procedure.

The detection of receptor-ligand interaction is generally accomplished by indirect assays such as the enzyme-linked immunosorbent and radio-labeled ligand assay. Although biotechnological functionalized films have led to  
10 elegant examples of molecular recognition at an interface, the problem of transducing the molecule recognition event into a measurable signal has remained a difficulty until the advent of the subject invention.

In the case of biosensor devices, detection is  
15 generally carried out by coupling the LB films to a secondary device such as an optical fiber (Beswick, Journal Colloid Interface Science, Vol. 124, p. 146, 1988), quartz oscillator (Furuki et al., Thin Solid Films, Vol. 210, p. 471, 1992), or electrode surfaces (Miyasaka, et al.,  
20 Chemical Letters, p. 627, 1990).

Some of the analytes bound films provide for fluorescent label, where the fluorescence or its quenched state indicate the occurrence of a binding event (Beswick, Journal Colloid Interface Science, Vol. 124, p. 146, 1988).  
25 In some cases, these detection materials have been embedded in the surface of the supporting bi-lipid layer (Tieke, Advanced Materials, Vol. 3, p. 532, 1991).

Polydiacetylene films are known to change color from blue to red with an increase in temperature or changes in pH  
30 due to conformational changes in the conjugated backbone (Ming, et al., Langmuir, Vol. 8, p. 594, 1992; Chance, et al., Journal of Chemistry and Physics, Vol. 71, p. 206, 1979; Shibutag, Thin Solid Films, Vol. 179, p. 433, 1989; Kaneko, et al., Thin Solid Films, Vol 210, p. 548, 1992).

Functionalized Liposomes

Unpolymerized liposomes expressing sialic acid residues have been extensively used as model systems to study the interaction between influenza virus and cell surfaces (Ott, 5 et al., European Journal of Pharmacological Science, Vol. 6, p, 333, 1994). These liposomes are typically made of such lipid materials as cholesterol and egg phosphatidylcholine (Kingery-Wood, et al., Journal of the American Chemical Society, Vol. 114, p. 7303, 1992).

10 In a publication which serves the basis for a U.S. Patent Application from which the subject application depends, is described a therapeutic fictionalized liposome which is produced through polymerization. The standard in the field is to progress with the polymerization procedure 15 until the materials are fully red, indicating that the polymerization is complete. This was the procedure used in the above cited publication.

While it has been a goal of the research community to exploit this characteristic in the detection of binding 20 events, researchers have yet to develop a method using this phenomenon in practical applications.

## GENERAL DESCRIPTION OF THE INVENTION

The present invention allows direct detection of small molecules, pathogens, bacteria, membrane receptors and 25 drugs, by the observation of color changes which occur when these analytes bind to the inventive three-dimensional polymeric assemblies. This technological advancement represents a dramatic improvement in results over the inventors 2-D prior monolayer film work, in that the color 30 intensity is dramatically improved. Additionally, the present work enjoys the many advantages which accrue when a test system can be suspended in fluid or bound to various supports.

It is an object of the present invention to assay the presence of biomolecules by directly detecting the binding event when the analytes specifically binds to three-dimensional polymeric assemblies.

5        It is a further object of the present invention to provide for the direct detection of viruses, bacteria, parasites, and other pathogens, and drugs, hormones, cell wall fragments, membrane fragments, membrane receptors, enzymes, and other biologically relevant materials using the  
10    inventive assay system.

It is another object of the present invention to provide for the development and improvement of drugs by observing competitive inhibition of natural binding events between all surfaces or binding sites and their natural  
15    bioactive ligand.

It is yet another object of the invention to provide means of testing libraries of materials, as the binding can be observed and the relevant liposome with its relevant ligand segregated from the others by segregating out a  
20    particular polymeric structure.

The present inventive assay means and method provide for the direct colorimetric detection of a receptor-ligand interaction using a novel three-dimensional polymeric assemblies system. Using the inventive method of producing  
25    these original assemblies, a ligand or its derivative is rendered polymeric by polymeric linking of the ligands through a linking arm, or through direct incorporation during the polymerization process. Some of these aspects of the present invention are described in the inventor's recently  
30    published communication, incorporated by reference herein, (Reichert et al., J. Am. Chem. Soc., Vol. 117, p. 829, 1995).



The presence of an analyte which binds to the incorporated ligands can be detected by observing changes in the spectral characteristics of the polymeric assemblies. The polymer-ligand assembly thus encompasses a molecular  
5 recognition site and a detection site, all within a single molecular assembly.

In one embodiment of the invention, chromatic polydiacetylene liposomes are produced, and placed in a liquid. The test sample is added. The color change which occurs indicates  
10 the presence of the analyte, and the intensity of the color allows a quantification of the analyte's concentration.

In the liposome embodiment of the present invention, the inventors have prepared synthetic, polymerizable liposomes that resemble the organization and  
15 functionalization of cell membranes and have employed them as simple colorimetric sensors. The liposomes were designed to specifically bind to influenza virus particles and, in addition, report the binding event by undergoing a visible color change. In effect, these molecular assemblies mimic  
20 cell surface molecular recognition as well as signal transduction.

In order to impart both molecular recognition and detection functions to the liposomes, the inventors combined a known ligand - receptor interaction with the unique  
25 optical properties of polydiacetylenes. The conjugated backbone of alternating double and triple bonds gives rise to intense absorptions in the visible spectrum. In single crystals or Langmuir-Blodgett films, these materials are known to undergo blue to red color transitions due to a  
30 variety of environmental perturbations including heat, mechanical stress, pH, and solvent.

In one embodiment of the subject invention, the inventors have demonstrated that specific binding of influenza virus to functionalized polydiacetylene liposomes

produces an analogous color transition. In earlier work, the inventors showed that similar effects can be obtained with functionalized 2-D polydiacetylene Langmuir-Blodgett films. (Charych, et al., Science, Vol. 261, p. 585, 1993).

5        ~~Influenza virus particles are enveloped by a lipid~~  
~~bilayer to which the hemagglutinin (HA) lectin is anchored.~~  
HA binds to terminal alpha glycosides of sialic acid on  
cell-surface glyco-proteins and glycolipids, initiating cell  
~~infection by the virus. As described in the prior art~~  
10 section of the subject application, liposomes expressing  
sialic acid residues have been extensively used as model  
systems to study the interaction between influenza virus and  
cell surfaces. The polymerized liposomes of the subject  
invention, however, are composed of molecules that allow  
15 direct visualization of this specific interaction.

#### Advantages of the Invention

Analytical Chemistry Techniques Analytical chemistry  
techniques are the only assay system prior to the advent of  
the subject invention that allows direct detection.  
20 Unfortunately, analytical chemistry have limited  
applicability to many biological system's assay needs.  
Unless expensive and cumbersome gas chromatography methods  
are used, large quantities of analyte are required. Often,  
quantitative results from such methods are limited or not  
25 available. However, such techniques have been used for such  
tests as hematocrit analysis, and creatinine assays.

Analytical chemistry methods are virtually unavailable  
for most biological molecules due to the destruction of the  
analytes characteristics during preparation and analysis  
30 steps, and the typically small amount of the analyte present  
in the test sample. For these reasons, the advent of  
immunoassay techniques were revolutionary in the biological  
sciences.

Immunoassays Many small biological molecules are notoriously difficult to assay in a direct manner due to the severe limitation of environmental ranges which they can tolerate without losing their specific characteristics. For these among other reasons, immunoassays have been heavily relied upon to assay these classes of materials. While successful in many respects, the indirect nature of immunoassay methods results in a variety of interferences, complications, problems, and assay limitations.

10 The requirement that an antibody be developed and produced for each possible target limits the efficacy of immunoassay methods in such applications as designer drug development and screening. Ironically, while allowing testing within a portion of biological environmental ranges, 15 the large glycoproteinaceous antibody are often highly sensitive to degradation outside of a small testing parameter environmental range. Thus, the susceptibilities of antibodies too rigorously limit the environmental testing range available in these assay systems.

20 A subtle disadvantage to immuneassay systems occurs in rapidly evolving pathogens such as the influenza virus. In such organisms, especially in the case of viruses, the external coat which is available for immune reactions has become constantly shifting in its antibody recognition 25 elements. Thus, despite a full blown immunity response to an influenza strain, within months an individual can again develop flu, but from a pathogen with an external coat so modified that it is immunologically unrecognizable by the victims memory cells. This is the reason individuals can 30 develop flu year after year.

#### Unique Qualities of The Present Invention

The present invention enjoys the unique advantage over both immunumoassay and analytical chemistry techniques of directly detecting biological analytes. In contrast to

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assays requiring binding to immunoglobulins, in one embodiment of the present invention, the host attachment site on the pathogen is exploited for recognition function. This site, generally in an immunologically inaccessible valley on the pathogen surface, is highly genetically conserved over time. ~~The minimal variability of this site~~ is necessary for the pathogen to maintain its infectivity. As a result, a single assay system of the present invention will provide effective assays for a panoply of influenza strains, many of which may be very newly evolved.

There are many advantages to the genetically conserved host recognition site being targeted by the embodiment of the present invention. A determination of a patient's exposure to the flu will be definitive, and not limited to a particular strain. This advantage of the present invention also avoids the need for a large number of immunological tests, as the clinician can rely on a single assay. Additionally, even newly evolved, uncharacterized flu strains can be identified, further avoiding false negative tests.

An analogous limitation of immunoassays occurs in well established pathogens such as malaria parasites. In these organisms, phases of the life cycle which would allow for an immune response have over time been so limited as to avoid the immune response, or have been made to occur within host cells so as to avoid an antibody reaction.

The present invention exploits the genetically conservative host binding site to identify the pathogen. Even in comparatively large parasites, the host binding site tends to be held constant over time throughout the generations of pathogens. Additionally, parasites are usually present in the body in a large number of diverse life stages. In well established parasites, the immune accessible sites often vary considerably from stage to stage, the advantage being that the host organism is unable

to mount a immunological response with sufficient rapidity to avoid the entrenchment of the parasite.

General Advantages of the Invention The subject invention represents a dramatic advancement over both prior art direct chemical and immunoassay systems, achieving advantages which, prior to the present invention, were available exclusively in only one or the other of these analytic art methods. Much as the advent of immunoassay techniques revolutionized medical and research analytical capacities, the subject invention represents a critical advance in the analytical arts.

The present invention allows the advantages of both immunoassay and chemical analysis in a single system. The present invention enjoys the direct assay advantages of analytical chemistry methods, with many of the advantages inherent in such systems. The inventive assay technique also has a substantial environmental range of testing beyond that of immunoassays. This allows the accommodation of various analyses in their most advantageous environmental parameters. Additionally, the present invention allows rigorous, direct analysis to occur even in very narrow environmental ranges, previously unavailable with analytical chemistry techniques. The speed and simplicity of the color change indicator of the subject invention are its hallmark advantages.

#### Target Materials

One of the unique advantages of the subject invention is the wide range of target materials, binding events, and biochemical reactions amenable to analysis using the inventive techniques. Many of these materials previously could not be detected using a straightforward, practical assay. The present invention allows many advantages of immunoassay systems, without the complications of immunoglobulin generation or indirect analysis.

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In general, the present invention requires no preanalysis purification step. This feature of the subject invention is due to the high specificity of the ligands incorporated into the detecting polymeric assembly.

5 Additionally, the inventive direct assay system avoids the expense, complications, and increased inaccuracies inherent in the indirect systems currently available.

Sensitive Analytes-Gentle Testing Conditions The inventive polymeric assemblies can employ ligands and

10 analytes which are stable or enjoy appropriate binding characteristics a limited in vitro or environmental range of conditions. Within in vitro range conditions, the present invention is useful in that stringent limitations even within this narrow range of conditions can be met. This

15 allows, for instance, three dimensional conformations of sensitive biochemicals and biomolecules to be maintained throughout the testing procedure.

The present invention functions well even in carefully limited conditions. Thus, conditions such as pH, saline,

20 and temperature can be carefully controlled by feedback controls, titration and other techniques without interfering with the accuracy or sensitivity of the analysis.

Because of this wide experimental range advantage of the present invention, intact cells or sensitive subcellular

25 inclusions can be assayed without disturbing their structural integrity. The color change when the inventive assemblies bind to a surface will pinpoint the location of an analyte, such as in a tissue sample.

Subtle cellular development stages can be monitored by

30 the present invention, such as the various stages of malaria infection. Additionally, the association between various factors can be tested or monitored even during the interaction process using the method of the subject invention.

Weak Binding Analytes-Multivalency The multivalent feature of the polymer-linked ligands of the subject invention provide a heightened binding capacity in the case of naturally multivalent analytes. Multivalency can also be provided for limited valency analytes prior to the test procedure to imbue them with this advantage of the subject invention. The inventive exploitation of multivalency allows a specific but weak interaction to be amplified many fold.

10 A structural linker of sufficient length and conformability aids in allowing binding of multiple sites on the analyte even when they are conformationally separated on a curved surface. As a result of these special features, the present invention can detect many ligands previously  
15 unsuitable for assay evaluation.

The main criteria for effective indication of the presence of analyte is that the surface of the polymeric assemblies be sufficiently perturbed to produce the requisite spectral change. Binding the analyte to an immobilizing particle will serve this purpose, as it  
20 concentrates the analyte in a small area, and further provides a three-dimensional aspect over a relatively large area to even a small analyte.

A large variety of ligands can be employed in the subject invention, allowing great flexibility in detecting  
25 a multivalent test target. Ligand selection can be based on the most advantageous binding and steric characteristics, rather than compromising these factors to accommodate the test system. Thus, the most advantageous ligand can be selected based on such factors as hydrophobicity and  
30 hydrophilicity, size, position of binding site, and conflicting affinities. Ligands which can be employed in the subject invention can include carbohydrates, peptides, nucleotides, heterocyclic compounds, and other organic  
35 molecules.

Challenging Analytes The rigor and outstanding advantages of the inventive assay system allows the detection and quantitative evaluation of materials which have been previously unachievable because of the limitations  
5 of the prior art methods.

The inventive construct and method can assay very small biological or other molecules for which antibodies can not be developed. These target materials can include organic solvents or pollutants present at extremely low levels.  
10 There are special opportunities made available by the advances achieved by the subject inventors for drug screening in both forensic and clinical applications. Inhibition techniques applied to the subject invention can allow the testing of materials which are of a tiny size or  
15 have a small number or single valiancy.

While applicants are not bound there by, it is hypothesized by the inventors that the unexpected spectral signal achieved by the present invention is due to a physical perturbation of the polymeric assemblies which  
20 occurs as a result of the binding event. It is the case that multivalent materials, such as viruses and cell membrane fragments, can be very easily detected using the subject inventive method. Thus, multivalent materials generally elicit a particularly strong response in the  
25 subject system. This may be the case because of conformational changes introduced into the lipid bi-layer as a result of binding causing physical reconfiguration of structure.

If applicants theory holds true, pre-binding of  
30 smaller, single valent analyte materials to a carrier may prove advantageous to increasing the efficacy of the subject invention in those cases. For instance, the analyte could be bound to a polymer or the surface of a liposome. This would concentrate the binding event on the inventive  
35 polymeric assemblies surfaces to specific points, increasing



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the spectral modification at each point of contact. Additionally, the curved surface of the liposome to which the analyte is attached will likely serve to tug the peripheral bound analyses away from the bilipid surface and  
5 force analytes centrally located on the liposome into the bilipid surface. This pre-binding step then can result in increased torsion, perturbation and signal generation on the bilayer surface.

### Signal Observation

10 Various spectral changes to the bi-layer can be used to detect the presence or absence of the target material. Means of amplifying the spectral signal well known in the art, such as scintillators, can also be employed when low levels of analyte are present. Because of the empirical  
15 nature of the signal, there are many opportunities for automating the read out of the present inventive assay system.

In one particular embodiment of the present invention, a blue pink color shift can be observed simply by visual  
20 observation by the testing technician. Because of the simplicity of the observation, this function can easily be accomplished by an untrained observer such as an at-home user. Alternatively, spectral test equipment well known in the art can be employed to determine a change in spectral  
25 qualities beyond the limits of simple visual observation, including optical density to a particular illuminating light wavelength.

The subject assemblage can also be optimized in assays by binding them to any one of a number of immobilizing  
30 materials and objects. Bonding to sephedex beads, for instance, would allow flow-through and washes to be possible during the assay procedures. The inventive assemblies could even be embedded in a gel, with the analyte diffusing through it, possibly with an electrical gradient.

## BRIEF DESCRIPTION OF THE DRAWINGS

Figure 1 shows a bifunctional molecule and a pentacosadiynoic acid.

Figure 2 is a color photograph showing a suspension of the inventive liposomes before and after introducing analyte.

## DETAILED DESCRIPTION OF THE INVENTION

The inventive three-dimensional polymeric assemblage allow for the direct detection of the presence of a wide range of analytes by changes in color. The results can be read by an untrained observer, and the test can be conducted in ambient conditions. Very mild testing conditions are possible, which allows the detection of small biomolecules in a near natural state, providing information as to their interactions and avoiding the risk of modification or degradation of the analyte.

The inventive polymer assemblies are composed of a three-dimensional structure, such as a liposome or tubule whose surface contains both orienting and detecting head groups. The detecting head groups are composed of a ligand specific to the analyte in question, which is bound to one terminal end of a linear structural linker. This linker, in turn, is bound to the polymeric assemblies by its second terminal end. The polymeric assembly surface is also provided with lipid ordering head groups.

Figure 1 provides a schematic depiction of one embodiment of the present invention. Receptor-binding ligand 1 is shown attached to one terminal end of spacer molecule 3. The second terminal end of spacer molecule 3 is then attached to one of several monomers which have been polymerized into a chromatic detection element 5. These materials are then agitated while polymerization occurs,

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causing the formation of the polymeric structures, such as liposomes and tubules.

Lipid Ordering Groups The lipids appear to be important in structurally ordering the three-dimensional polymeric assemblies so that binding of the analyte produces a detectable color change. Applicants hypothesize that a structuring effect of the ordering groups serves to appropriately stabilize the physical structure of the three-dimensional polymeric assemblies to facilitate color stability and polymerization. In turn, the binding of the analyte to the molecular recognition ligand groups then causes sufficient steric perturbation or stress of the structure to result in a color change. It may be that the stability and relative rigidity engendered by the ordering lipids so unites the bilayer surface, that a steric change in one area triggers a larger effect in the surface of the assemblies as a whole.

It is not certain which of the many results of binding result in the observed spectral changes. Most likely the changes are due to stresses induced by binding which changes the effective conjugation length of the polymer backbone. The inventive three-dimensional structures are highly color sensitive to a number of environmental parameters, such as heat, and these factors may be a component of the observed phenomena as well. However, the applicants are not bound to any of the above hypothesis which are simply attempts to explain the demonstrated subject invention.

Previous studies have suggested that color transitions in polydiacetylenes arise from changes in the effective conjugation length of the polydiacetylene backbone and that effective assay method of the electronic structure of the polymer backbone is strongly coupled to side chain conformation. The inventors can only speculate at this point that specific virus-liposome interactions may serve to alter side chain conformation, reducing the effective

conjugation length of the enzyme backbone. Indeed, theoretical calculations suggest that very slight around the C-C bond of the polymer backbone decrease the  $\pi$  electron delocalization.

- 5           Materials for use are as head groups in the present invention include  $-\text{CH}_2\text{OH}$ ,  $-\text{CH}_2\text{OCONHPh}$ ,  $-\text{CH}_2\text{OCONHEt}$ ,  $-\text{CH}_2\text{CH(Et)OCONHPh}$ ,  $-(\text{CH}_2)_9\text{OH}$ ,  $-\text{CH}_2\text{OCOPh}$ ,  $-\text{CH}_2\text{OCONHMe}$ ,  $-\text{CH}_2\text{OTs}$ ,  $-\text{CH(OH)Me}$ ,

- $-\text{CH}_2\text{OCOR}_2$ , wherein  $R_2$  is  $n\text{-C}_5\text{H}_{11}$ ,  $n\text{-C}_7\text{H}_{15}$ ,  $n\text{-C}_9\text{H}_{19}$ ,  $n\text{-C}_{11}\text{H}_{23}$ ,  $n\text{-C}_{13}\text{H}_{27}$ ,  $n\text{-C}_{15}\text{H}_{31}$ ,  $n\text{-C}_{17}\text{H}_{35}$ ,  $\text{Ph}$ ,  $\text{PhO}$ , or  $o\text{-(HO}_2\text{C)C}_6\text{H}_4$ ,

$-\text{OSO}_2\text{R}_2$ , wherein  $R_2$  is  $\text{Ph}$ ,  $p\text{-MeC}_6\text{H}_4$ ,  $p\text{-FC}_6\text{H}_4$ ,  $p\text{-ClC}_6\text{H}_4$ ,  $p\text{BrC}_6\text{H}_4$ ,  $p\text{-MeOC}_6\text{H}_4$ ,  $m\text{-CF}_3\text{C}_6\text{H}_4$ ,  $2\text{-C}_{10}\text{H}_7$ , or  $\text{Me}$  -  $\text{CO}_2\text{M}$ , wherein  $\text{M}$  is  $\text{K}$ ,  $\text{HNa}$ , or  $\text{Ba}/2$ .

- 15           The preferred materials which can be employed as head groups in the present invention are:

$-\text{CH}_2\text{OCONHR}_2$  or  $-\text{CH}_2\text{CONHR}_2$  where  $R_2$  is  $\text{Et}$ ,  $n\text{-Bu}$ ,  $n\text{-C}_6\text{H}_{13}$ ,  $n\text{-C}_8\text{H}_{17}$ ,  $n\text{-C}_{12}\text{H}_{25}$ ,  $\text{cyclo C}_6\text{H}_{11}$ ,  $\text{Ph}$ ,  $p\text{-MeC}_6\text{H}_4$ ,  $m\text{-MeC}_6\text{H}_4$ ,  $o\text{-ClC}_6\text{H}_4$ ,  $m\text{-ClC}_6\text{H}_4$ ,  $p\text{-ClC}_6\text{H}_4$ ,  $o\text{-MeOC}_6\text{H}_4$ ,  $3\text{-Thienyl}$ ,  $\text{Me}$ ,  $\text{Et}$ ,  $\text{Ph}$ ,  $1\text{-C}_{10}\text{H}_7$ ,   
20  $\text{Et}$ ,  $\text{Ph}$ ,  $\text{EtOCOCH}_2$ ,  $\text{BuOCOCH}_2$ ,  $\text{Me}$ ,  $\text{Et}$ ,  $i\text{-Pr}$ ,  $n\text{-C}_6\text{H}_{13}$ ,  $\text{EtOCOCH}_2$ ,  $\text{BuOCOCH}_2$ ,  $\text{Ph}$ ,  $2,4(\text{NO}_2)_2\text{C}_6\text{H}_3\text{OCH}_2$ , or  $\text{CH}_2\text{CH}_2\text{OH}$ .

The most preferred head groups are taken from  $-\text{CH}_2\text{COX}$ , where  $\text{X}$  is  $\text{OH}$ ,  $\text{MeO}$  or any salt thereof.

- Ligand Group The ligand group of the present invention  
25 can be of a wide variety of materials. The main criteria is that the ligand have an affinity for the analyte of choice. The ligand may be of a broad range, such as when a class of materials is to be assayed. Appropriate ligands include peptides, carbohydrates, nucleic acids or any organic  
30 molecules which bind to receptors. For instance, all influenza strains share binding sites to a host receptor molecule. Thus, this molecule can successfully be employed to screen for all influenza strains, including those which have not yet been characterized.

Ligands can also be used in the present invention when they function as competitive binders to the analyte. For instance, a pathogen could be introduced with a test material which is to be the presence of receptor molecule. In absence of this molecule, the pathogen will bind to the three-dimensional polymeric structure and produce a color. To the degree that the pathogen surface is bound to the receptor molecule introduced in the test material, the binding will be diminished. In this way, the presence of receptor molecule can be detected and quantified.

Receptor-Binding Molecules The use of sialic acid derivatives in one preferred embodiment described in the examples below is an example of the use of receptor-binding molecules in this capacity. Receptor-binding molecules are materials on the surface of a host cell to which a pathogen attaches itself as a prelude to the infective event. Selecting these molecules at the ligand group in the present invention has many advantages over other receptor molecules.

The recognition site for these molecules tend to be highly genetically conserved in the pathogen because of its obvious criticality to survival. Therefore, different strains of the same pathogen will generally not produce a false negative when such molecules are selected as the ligand group in the subject invention. Also, receptor molecules tend to be smaller and less complex, and often less hydrophobic, than antibodies to the same analyte.

An increasing number of receptor molecules are being recognized, identified, isolated, and synthesized for a large number of pathogens. Many have been improved for use in various analytic and treatment systems. An example of this trend in research is the sialic acid derivative used in the example below of the subject invention. Examples of the receptors for a number of pathogens are provided in the application as Table 3. All of these, as well as many more, could be exploited by the method of the subject invention.

TABLE 3

Pathogen	Receptor Molecule	CD4 <sup>+</sup> ; Vasoactive Intestinal Peptide <sup>1</sup> , Peptide T <sup>2</sup> , Sialic Acid <sup>13</sup>
HIV	Epidermal Growth Factor <sup>1</sup>	
Vaccinia	Acetylcholine receptor <sup>3</sup>	
Rabies	Complement Receptor <sup>4,5</sup>	
Epstein Barr	Beta-adrenergic receptor <sup>6</sup>	
Rheo	ICAM-1 <sup>10,11</sup> , N-CAM, myelin-associated glycoprotein (MAG) <sup>12</sup>	
Rhinovirus	Polio virus receptor <sup>7</sup>	
Polio viruses	Sialic Acid <sup>15</sup>	
Influenza	Glycoprotein (not Sialic Acid) <sup>16,17,18</sup>	
Cytomegalovirus	9-OAC Sialic Acid & Sialic Acid	
Coronaviruses	9-OAC Sialic Acid	
Encephalomyelitis	----- <sup>19</sup>	
Rubella Virus	Glycoprotein (not Sialic Acid) <sup>20,21,22,23</sup>	
Measles Virus	Oligosaccharides Glycoprotein <sup>24,25,26</sup>	
Herpes	Sialic Acid <sup>27,28,29,30</sup>	
Chlamydia	Glycosylated Proteins <sup>31,32</sup>	
Rhinovirus	9-OAC Sialic Acid	
Rotavirus	Sialic Acid	
Polyomavirus	Sialic Acid	
Reovirus	Sialic Acid	
Streptococcus Suis	Sialic Acid $\alpha$ 2 $\rightarrow$ 3 Poly-N-Acetylactosamine	
Salmonella	Sialic Acid	
Typhimurium	Sialic Acid	
Paramyxovirus	Sialic Acid	
Sendi Virus	Sialic Acid	
Mumps	Sialic Acid	
Newcastle	Sialic Acid	
Disease Virus	Sialic Acid	
Myxoviruses	Oligomannose, Galactose $\alpha$ 1 $\rightarrow$ 4 Galactose, Sialic Acid $\alpha$ 2 $\rightarrow$ 3 Galactose	
Escherichia Coli	Sialic Acid	
Encephalomyocarditis	G <sub>M1</sub> (A Gangliosial of Sialic Acid, Galactose, Glucose, N-Acetyl Galactose)	
Virus	Sialic Acid	
Cholera Toxin		
Meningitis		

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Lipid Polymerization Groups Many different polymerizing groups have been incorporated into lipids and are shown to be effective in monolayer polymerizations. Such moieties include: acetylenes, diacetylenes, alkenes, thiophenes, imides, acrylamides, methacrylates, vinyl ether, malic anhydride, urethanes, allylamines, siloxanes or vinylpyridinium etc. Lipids containing these groups can be made into homopolymers or mixed polymers. The preferred group for use in this invention is the diacetylene due to its unique optical properties in the polymerized form: polydiacetylene. However, other polymerizing groups could be used when they provide an observable change in properties upon a binding event.

Forms of the Assemblies The three-dimensional assemblies of the subject invention can be produced in any number of forms. One of the most important forms which can be produced are liposomes. Several methods for producing the subject assemblies into that particular form are fully set forth in the Example section of this application.

The liposomes of the subject invention can be formed in a number of different sizes and types. For instance, it is possible to form the liposomes as simple bi-layer structures. Additionally, they can be multi-layered, in an onion type structure. Their size can also be varied.

Numerous other shapes can also be produced. Double chains (Kuo et al, Macromolecule, p. 3225, Vol. 23 1990), lamellae (Rhodes, et al, Langmuir, p. 267, Vol. 10, 1994), hollow tubules and braids (Franker et al, Journal of the American Chemistry Society, Vol. 116, 1994), among other shapes can be formed. When these assemblies are immobilized, they can collectively form even larger constructs.

One example of a successful protocol for producing the liposome embodiment of the subject invention is as follows:

- mixing of the appropriate amounts of the chloroform solutions of the lipids (1-15mM) in a small vial
- evaporation of the chloroform with a stream of nitrogen
- addition of the appropriate amount of de-ionized water (total lipid concentration 1-2mM)
- heating of the solution above the phase transition of the lipids (about 80-90°C)
- ~~sonication of the solution for 15 minutes~~ (probe sonicator, Fisher sonic dismembrator model 300, 50% maximum output, microtip)
- filtration of the warm opaque solution through a 0.8  $\mu$ m nylon filter (Gelman) to remove small titanium particles from the solution
- cooling of the solution for at least one hour up to one day in the fridge (4°C)
- removal of the oxygen in the solution by bubbling nitrogen through the sample for 5-10 minutes prior to polymerization
- polymerization of the stirred liposome solution in a 1 cm quartz cuvette with a small 254 nm UV-lamp (pen-ray, energy: 1600 $\mu$ w/cm<sup>2</sup>) in a distance of 3 cm in a small chamber which is purged with nitrogen 20 minutes prior to and during the polymerization to replace all oxygen and to cool the sample; polymerization times vary between 5 and 30 minutes depending on the desired properties (color, polymerization degree) of the liposomes.

#### EXAMPLE 1

As shown in Fig. 1, the bifunctional molecule 1 use in one embodiment of the subject invention incorporates both the sialic acid ligand for viral binding and the diacetylenic functionality in the hydrocarbon chain for polymerization. The carbon-glycoside in this compound was



-22-

designed to prevent hydrolysis by viral neuraminidase. This compound was mixed with 10-12-pentacosadiynoic acid 2 and hydrated to form liposomes. Although most natural lipids that form liposomes consist of two alkyl chains, synthetic  
5 liposome-forming lipids with only one alkyl chain also exist. See, for example: Hunfer et al Phys. Lipids, pp. 355-374 Vol. 33, and Bader, Chem., Int. Ed. Engl., pp 91-92 Vol. 20, 1981. Previous studies indicated that optimum viral binding occurs for mixtures of 1-10% compound 1 in the  
10 liposome. Spevak et al, J. Am. Chem. Soc., 161, 115 & p. 1146, 1993. Therefore, 5% and 10% sialic acid lipid were used in this colorimetric detection study.

Liposomes were prepared using a probe sonication method, [Liposomes: A Practical Approach; New, Ed.; Oxford  
15 University Press; Oxford, pp 33-104, 1990,] and subsequently an ultraviolet polymerized by irradiation at 254 nm using pencil lamp. Table 1 shows colorimetric detection of liposomes (5% influenza virus by polymerized diacetylene sialic acid lipid 1), including the visible absorption  
20 spectra of (A) blue liposome solution (8 min W) and (B) purple liposome solution (24 min W) without virus (solid line) and after incubation with 60 HAU of influenza virus (dashed line). The concentration of the liposome solutions in PBS buffer was 0.13 mM, and the incubation time with the  
25 virus was 1h.

Irradiation of a liposome solution (1mM in deionized water) for about 5-10 min results in the formation of deeply blue colored liposomes (Table 1A, solid line). If the polymerization time is longer (between 10 and 30 min), a  
30 purple color is observed (Table 1B, solid line). When influenza virus is added to the liposomes in PBS buffer, the solution immediately changes to a pink or orange color, depending on whether the initial preparation was blue or purple, respectively (Table 1A and B, dashed curves). These  
35 color changes are readily visible with the naked eye and can be quantified by visible absorption spectroscopy.

22 (a)

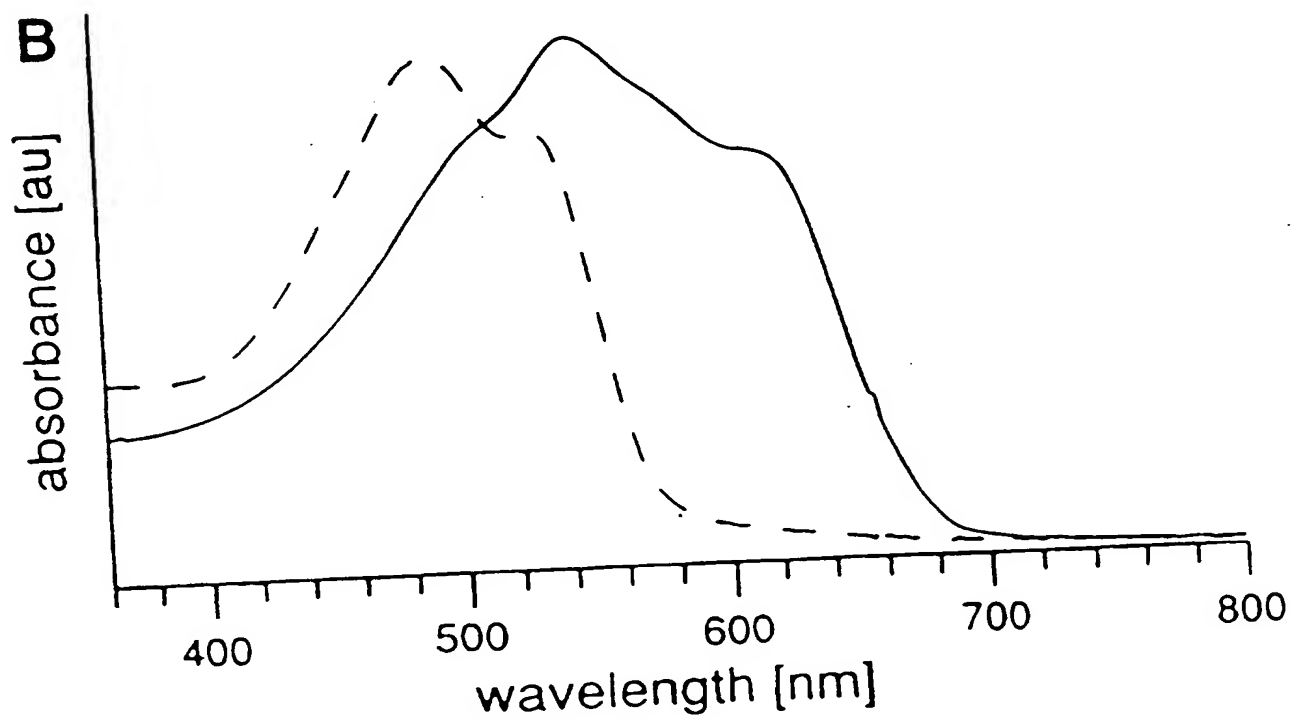
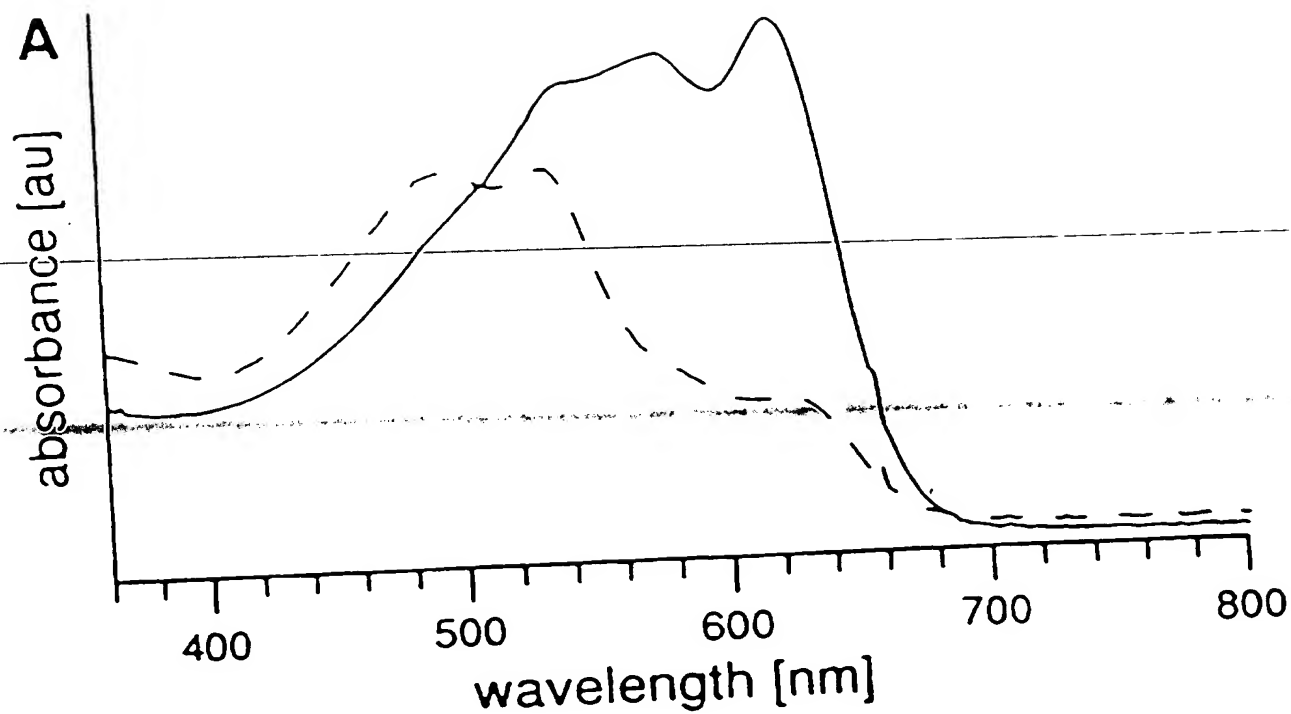


Table 1

The colorimetric response (CR) is quantified by measuring the percent change in the absorption at 626 nm (which imparts the blue color to the material) relative to the total absorption maxima. In order to quantify the response of a liposome solution to a given amount of virus, the visible absorption spectrum of the liposome solution without the virus was analyzed as

$$B_o = I_{626} / (I_{536} + I_{626})$$

where  $B_o$  is defined as the intensity of absorption at 626 nm divided by the sum of the absorption intensities at 536 and 626 nm. The liposome solution which was exposed to influenza virus was analyzed in the same way as

$$B_v = I_{626} / (I_{536} + I_{626})$$

where  $B_v$  represents the new ratio of absorbance intensities after incubation with the virus; The colorimetric response (CR) of a liposome solution is defined as the percentage change in B upon exposure to virus

$$CR = [(B_o - B_v) / B_o] \times 100\%$$

To be consistent with the inventors' earlier work, the absorption maxima at 626 and 536 nm was arbitrarily chosen to calculate the percentage blue absorption for the liposomes solutions. Use of the second absorption maximum at 480 nm for the calculations does not change the relative trend of the results shown.

As shown in Table 1, incubation of the blue liposomes (8 min UV) with 60 hemagglutinating units (HAUs) of virus leads to a CR of 47%; incubation of the purple liposomes (24 min UV) with the same amount of virus gives a CR of 87%. A hemagglutinating unit (HAU) is a measure of the highest dilution of the virus solution that still completely agglutinates a 1% solution of red blood cells. The inventors speculate that the enhanced sensitivity of the purple liposomes may be due to an increased polymer content, as suggested by their higher optical density (data not shown).

No color change could be detected if pure PBS buffer or a solution of BSA in PBS buffer (1mg/mL) was added to the liposome solution ( $CR \leq 5\%$  within 2 h). In order to directly address the effects of nonspecific adsorption, liposomes  
5 were prepared without sialic acid lipid 1 in Fig. 1. Similarly, these liposomes did not change color after exposure to virus.

## EXAMPLE 2

The specific nature of the interaction between the  
10 influenza virus and the sialic acid liposomes was confirmed by a competitive inhibition experiment. Incubation of a liposome solution (10% sialic acid lipid 1) with 54 HAUs of influenza virus yields a CR of 31% for blue and 70% for purple liposomes. Performing the same experiment with a  
15 slight excess of a-O-methyl-neuramatic acid, a known inhibitor for influenza virus hemagglutination, results in no color change.

Kinetic experiments show that the color change induced by the addition of an aliquot of virus reaches a plateau  
20 after 30 min. although the change becomes apparent within 5 min. For a given polymerization time, the CR depends on the amount of added virus, as shown in Table 2. Table 2 is the plot of the colorimetric response of a purple liposome solution (5% sialic acid lipid 1, 24 min UV) versus  
25 successive additions of influenza virus. The liposomes were incubated for 30 min following each addition of virus, and the visible absorption spectrum was recorded. The CR for each virus concentration was obtained in three independent experiments.

30 Given that the color change of the liposomes in buffer without virus is less than 4% within 2 h, a CR of 5% or more in a few minutes is considered significant. Therefore, the amount of virus required to produce a CR just above this value defines the detection limit of the method in this

24 (a)

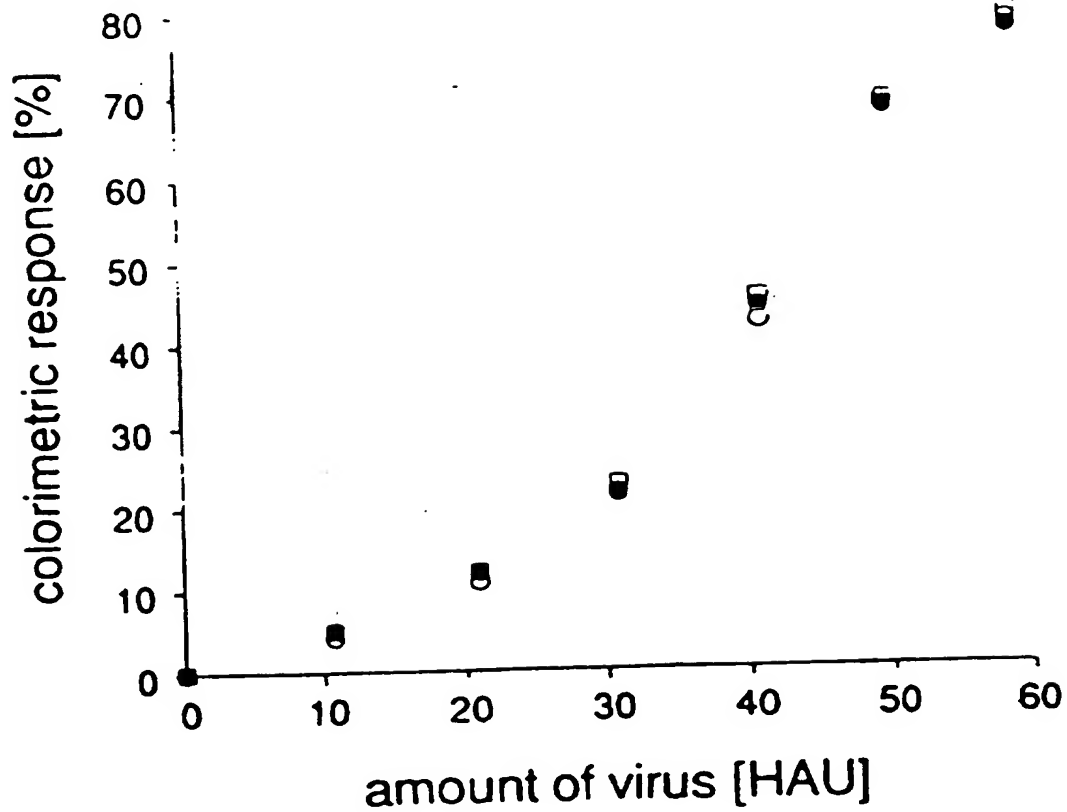


Table 2

particular embodiment. The titration curve in Table 2 shows that as little as 11 HAU's can be detected. This corresponds to approximately  $11 \times 10^7$  virus particles by electron microscopy count.

5        The subject inventors have demonstrated that  
polymerized structures including liposomes are biomolecular  
materials that provide a molecular recognition function  
(sialic acid) and a detection element (polydiacetylene  
backbone), all within a single supramolecular assembly. The  
10 binding event is transduced to a visible color change,  
readily seen with the naked eye and quantified by absorption  
spectroscopy. Specificity of the color change was  
demonstrated by competitive inhibition studies. In  
addition, nonspecific adsorption, if it occurs, does not  
15 appear to affect the color of the liposome solutions.

#### Example 3: Immobilizing Liposomes to Substrates

Attachment to membranes of poly(ether urethanes) or  
polyacrylonitrile. These membranes are porous, hydrophilic  
and can be used for affinity separations or immunodiagnosis.

20        The liposomes can be coupled to these membranes by  
first attaching to the membrane an activating group such as  
imidizolyl-carbonyl, succinimido, FMP or isocyanate which  
adds rapidly to nucleophiles in the liposomes such as -NH<sub>2</sub>,  
SH, -OH. Thus, any liposome preparation which contains  
25 these functionalities can be directly attached to the  
membrane. This procedure is analogous to the coupling of  
proteins to membranes the latter of which can be found in  
the literature. (C.H. Bamford, K.G. Al-Iamee, M.D.  
Purbrick, T.J. Wear, J. Chromatography, 1992, 606, 19 or  
30 C.H. Bamford, K.G. Allamee, Clinical Materials, 1992, 10,  
243. In principle, any strategy previously developed to  
immobilize proteins can be used to immobilize liposomes.

Liposomes which have an -SH functionality can also be immobilized directly to gold surfaces, particles, or electrodes via the thiol-gold bond. In this case, a solution of the liposomes containing the -SH group are  
5 incubated with the clean gold surface in water for 12 - 24 hours with stirring at room temperature.

Liposomes can be immobilized to silicon chips or silica gel (silicon dioxide) using the following procedure. The gel or wafers are acid cleaned in 1:1 HCl, -methanol, rinsed  
10 in water, and placed in concentrated sulfuric acid. After a thorough water rinse, the wafer chips or gel is boiled in doubly distilled deionized water, allowed to cool and dry and then silanized under inert atmosphere in a 2% solution of 3-mercaptopropyl trimethoxysilane prepared in dry  
15 toluene. Next, the chips or gels are placed in a 2 mM solution of either GMBS (N-succinimidyl 4-maleimidobutyrates) or EMCS (N-succinimidyl 6-maleimidocaproate) prepared in 0.1 M phosphate buffer (the cross linker is first dissolved in a minimal amount of dimethylformamide). After rinsing with  
20 phosphate buffer, the chips are placed in a 0.05 mg/ml solution of the liposomes prepared in pH 8.0 phosphate buffer. Finally, the chips or gels are thoroughly rinsed with, and then stored in, the buffer solution prior to their use. The liposomes should have an -NH<sub>2</sub> functionality for  
25 the cross-linking with GMBS or EMCS to work. This procedure is a modification of a previously developed procedure which was used to immobilize enzymes to silicon chips or gels. It has been modified for the liposome immobilization. (from K.M. Rusin, T.L. Fare, I.Z. Stemple, Biosensors and  
30 Bioelectronics, 1992, 7, 367).

-NH<sub>2</sub> functionalized liposomes can also be immobilized onto surfaces by use of standard glutaraldehyde coupling reactions such as often used with the immobilization of proteins.

Example 4: Detection and Screening

The liposomes can be used to replace standard radiolabel assays for ligand-receptor screening. For example, if the ligand is an analog of dopamine (e.g. the compound "spiperone"), the ligand can be incorporated into polymerized liposomes (polymerized assemblies). If the membrane receptor for dopamine, such as the dopamine D-2 receptor is added to the spiperone-modified liposomes, a color change from blue to pink is observed. This can be monitored spectroscopically in a manner similar to the detection of viruses and bacteria. The effect can be inhibited by the addition compounds which bind as strongly or stronger than dopamine or spiperone. By using a 96-well plate format, 96 compounds which are analogs of dopamine can be screened as potential new drugs. This high throughput screening does not require the use of expensive radiolabelled compounds and does not have the associated health and safety problems.

Procedure: Dilute 20 - 50 uL of liposome solution which contain from 0.5% - 20% of the spiperone ligand in 100 - 200 uL, of an appropriately buffered medium. The solution will have a blue or purple color. The visible absorption spectrum of the sample can be recorded at this point. For detection study: add the dopamine D2 membrane receptor preparation in successive aliquots starting at 10 - 50 uL until 100 - 200 uL. The color change can be observed by eye or by recording of the visible absorption spectrum. For drug screening studies: Add the dopamine D2 membrane receptor preparation mixed with the new ligand or new drug compound. Allow for binding to occur by incubating at room temperature or at 37°C for 5 - 60 minutes. Add the inhibited membrane receptor preparation to the diluted liposome solution. If the solution turns pink, the new ligand or drug was ineffective. If the solution remains blue, the new ligand or drug was an effective binder to the receptor.



Example 5: Detection of Radioactive Metals.

The monomeric diynes can be polymerized by exposure to gamma irradiation. By incorporating a ligand which is a metal chelator, the monomeric form of the liposomes are exposed to a solution of radioactive metals. Upon binding of the metal to the chelator ligand, the emitted gamma irradiation serves to polymerize the liposomes. The solution changes from a whitish opaque solution (unpolymerized liposomes) to a deep blue or deep red solution of the polymers. The liposomes serve two purposes: 1) to detect the presence of the radioactive metals, and 2) to clean the solution of the radioactive metals. Step 2 is accomplished by simply filtering or centrifuging the metal-bound liposomes. This procedure can be referred to as "seen and cleaned" since the liposomes both detect and purify the radioactive metals from the surrounding environment.

Procedure: Prepare liposomes as described up until the point of UV irradiation. The monomeric liposomes will have an opaque, whitish appearance. For detection: Dilute 10 - 100 uL of liposomes in 50 - 200 uL of water or appropriate buffer. The liposomes will contain 0.5% - 20% of the chelator ligand. This can be done in a 96-well plate format. Add the environmental sample to be tested, 50 - 100 uL. Observe the formation of a blue to red color indicating the presence of gamma irradiation, and hence the radioactive metal. For large scale cleanup purposes, the liposomes can be immobilized onto large filtration units near the efflux of wastewater treatment areas, for example, as found at Superfund clean up sites or at other DOE facilities. The treated water passes over the filtration units. Any remaining radioactive metals in the water will be detected by a blue or red color on the filtration unit. At the same time, these metals will be cleared from the treated water such that the water can be returned to the environment or retested.

Example 6: Glucose Sensor

The liposomes are sensitive to pH. At high pH the liposomes are in the red state and at low pH the liposomes are in the blue state. The effect can be made reversible.

5 The liposomes can be used to detect small molecule analyses which in the presence of an appropriate enzyme or other metabolic cellular process changes the pH of its surrounding media. For example, in the detection of glucose. The liposomes are added to a media of sufficiently high pH to

10 put them in the red state. 10 - 50 uL of liposomes can be diluted with 50 - 200 uL of the appropriate media. The test sample is added, 10 - 100 uL. This can be done in a 96-well plate format. To the test sample is added 10 - 50 uL of glucose oxidase. If glucose is present, the glucose oxidase

15 will convert glucose to glucaraonic acid. This conversion will lower the pH of the solution, producing the blue state of the liposomes. This red to blue color change signifies the presence of glucose in the sample. The test can be done visually or quantitatively by measuring the visible

20 absorption spectrum.

CLAIMS

1. Three-dimensional polymeric multilayer assay assemblies which change color in the presence of analyte comprising:
  - 5 a) ligands having direct affinity for the analyte or which can function as competitive binders to the analyte,
  - b) linear structural linkers having two terminal ends, wherein said linkers are attached at  
10 their first terminal ends to said ligand moieties,
  - c) a conjugated polymer backbone to which said structural linkers are bound at their second terminal ends, and
  - 15 d) ordering head groups which are bound to the surface of the conjugated polymer backbone in positions not occupied by the structural linker.
2. The assemblies of Claim 1, wherein said assemblies are in the form of a liposome, a double-chain, braided, lamellar, helical, tubular, or fiber-like shape.
- 20 3. The assemblies of Claim 1, wherein the analyses are biomedical materials, pathogens, drugs, radioactive metals, or industrial materials.
4. The assemblies of Claim 3, wherein said biomedical materials are selected from the group comprising  
25 pathogens and cells infected by them, drugs, hormones, blood components, disease indicators, cell components, antibodies, lectins, enzymes, genetic material, and their metabolic derivatives.
5. The assemblies of Claim 3, wherein said pathogens are  
30 selected from the group comprising viruses, bacteria, parasites and other pathogens.

6. The assemblies of Claim 5, wherein said virus are selected from the group comprising influenza, cold, rubella, chicken pox, hepatitis A&B, herpes simplex, polio, small pox, plague, HIV, vaccinia, rabies, Epstein Barr, reovirus, rhinovirus, and mutations, ligand recognizable parts thereof.
7. The assemblies of Claim 5, wherein said bacteria are selected from the group comprising E. coli, tuberculosis, salmonella, streptococcus, and mutations, strains and degraded parts thereof.
8. The assemblies of Claim 5, wherein said parasites and other pathogens are selected from the group comprising malaria, sleeping sickness, river blindness, and toxoplasmosis.
9. The assemblies of Claim 1, wherein said ligand is provided for the detection of a pathogen analyte.
10. The assemblies of Claim 9, wherein the analyte is a virus.
11. The assemblies of Claim 9, where said ligand is selected from the group comprising, epidermal growth factor for vaccinia analyte, acetylcholine receptor for rabies analyte, complement receptor for Epstein Barr analyte, beta-adrenergic receptor for reovirus analyte, ICAM-1 for Rhinovirus analyte, polio virus receptor for polio virus analyte, trisaccharide analyte for cholera toxin analyte, tetrasaccharide for neutrophil analyte, and derivatives and analogues thereof capable of associating with an analyte.
12. The assemblies of Claim 9, wherein said ligand is sialic acid and its derivatives and analogs which will bind to colronaviruses, influenza virus, encephalomyelitis, chlamydia, sendi virus, mumps,

newcastle disease, myxovirus, encephalo-mycarditis virus, meningitis, or malaria.

13. The assay of Claim 9, wherein the ligand: analyte pair are tetrasaccharides and neutrophils, cell adhesion peptides and target cells, trisaccharides and bacterial toxins or transmembrane receptors and hormones.
14. The assemblies of Claim 9, wherein the ligand provided to detect HIV analytes is selected from the group comprising CD4, sCD4, CD26, vasoactive intestinal peptide, peptide T, sialic acid, and derivatives and analogues thereof capable of associating with HIV.
15. The assemblies of Claim 1, wherein said polymer is comprised of polymerizable lipid monomers.
16. The assemblies of Claim 15, wherein said monomers are chosen from the group comprising acetylenes, diacetylenes, alkenes, thiophenes, imides, acrylamides, methacrylates, vinyl ether, malic anhydride, urethanes, allylamines, siloxanes, anilines, pyrroles and vinylpyridinium.
17. The assemblies of Claim 16, wherein said polymer backbone is comprised of diacetylene monomers.
18. The assemblies of Claim 1, wherein said ordering head groups are hydrophilic, with the capacity to mutually hydrogen bond.
19. The assemblies of Claim 18, wherein said ordering head groups are selected from the group comprising;  $-\text{CH}_2\text{OH}$ ,  $-\text{CH}_2\text{OCONHPh}$ ,  $-\text{CH}_2\text{OCONHEt}$ ,  $-\text{CH}_2\text{CH(Et)OCONHPh}$ ,  $-(\text{CH}_2)_9\text{OH}$ ,  $-\text{CH}_2\text{OCOPh}$ ,  $-\text{CH}_2\text{OCONHMe}$ ,  $-\text{CH}_2\text{OTs}$ ,  $-\text{CH(OH)Me}$ ,  $-\text{CH}_2\text{OCOR}_2$ , wherein  $\text{R}_2$  is  $n\text{-C}_5\text{H}_{11}$ ,  $n\text{-C}_7\text{H}_{15}$ ,  $n\text{-C}_9\text{H}_{19}$ ,  $n\text{-C}_{11}\text{H}_{23}$ ,  $n\text{-C}_{13}\text{H}_{27}$ ,  $n\text{-C}_{15}\text{H}_{31}$ ,  $n\text{-C}_{17}\text{H}_{35}$ ,  $\text{Ph}$ ,  $\text{PhO}$ , or

o-(HO<sub>2</sub>C)C<sub>6</sub>H<sub>4</sub>, -OSO<sub>2</sub>R<sub>2</sub>, wherein R<sub>2</sub> is Ph, p-MeC<sub>6</sub>H<sub>4</sub>, p-FC<sub>6</sub>H<sub>4</sub>, p-ClC<sub>6</sub>H<sub>4</sub>, pBrC<sub>6</sub>H<sub>4</sub>, p-MeOC<sub>6</sub>H<sub>4</sub>, m-CF<sub>3</sub>C<sub>6</sub>H<sub>4</sub>, 2-C<sub>10</sub>H<sub>7</sub>, or Me-CO<sub>2</sub>M, wherein M is K, H, or B a/2,

- 5                    -CH<sub>2</sub>OCONHR<sub>2</sub> or -CH<sub>2</sub>CONHR<sub>2</sub> where R<sub>2</sub> is Et, n-Bu, n-C<sub>6</sub>H<sub>13</sub>, n-C<sub>8</sub>H<sub>17</sub>, n C<sub>12</sub>H<sub>25</sub>, cyclo C<sub>6</sub>H<sub>11</sub>, Ph, p-MeC<sub>6</sub>H<sub>4</sub>, m-MeC<sub>6</sub>H<sub>4</sub>, o-ClC<sub>6</sub>H<sub>4</sub>, m-ClC<sub>6</sub>H<sub>4</sub>, p-ClC<sub>6</sub>H<sub>4</sub>, o-MeOC<sub>6</sub>H<sub>4</sub>, 3-Thienyl, Me, Et, Ph, 1-C<sub>10</sub>H<sub>7</sub>, Et, Ph, EtOCOCH<sub>2</sub>, BuOCOCH<sub>2</sub>, Me, Et, i-Pr, n-C<sub>6</sub>H<sub>13</sub>, EtOCOCH<sub>2</sub>, BuOCOCH<sub>2</sub>, Ph, 2,4(NO<sub>2</sub>)<sub>2</sub>C<sub>6</sub>H<sub>3</sub>OCH<sub>2</sub>, or CH<sub>2</sub>CH<sub>2</sub>OH.

10    20. The assemblies of Claim 18, wherein said ordering head group is a carboxylic acid.

21. The assemblies of Claim 1, wherein the non-binding terminal of the monomers is selected from the group comprising; CH<sub>3</sub>-, CH<sub>3</sub>O-, neo-C<sub>5</sub>H<sub>11</sub>O-, Cyclo-C<sub>6</sub>H<sub>11</sub>O, 15    PhCH<sub>2</sub>O-, p-AcC<sub>6</sub>H<sub>4</sub>O-, p-BZC<sub>6</sub>H<sub>4</sub>O- p-BrC<sub>6</sub>H<sub>4</sub>cOCH<sub>2</sub>O- p-(PhCH=CHCO) C<sub>6</sub>H<sub>4</sub>O-, p- (PhCOCH=CH)C<sub>6</sub>H<sub>4</sub>O- oBzC<sub>6</sub>H<sub>4</sub>NH-, p-BzC<sub>6</sub>H<sub>4</sub>NH-, MeOCH<sub>2</sub>CH<sub>2</sub>NH-, n-C<sub>6</sub>H<sub>13</sub>NH-, and EtO-.

22. The assemblies of Claim 21, wherein the terminal is a methyl group.

20    23. The assemblies of Claim 1, wherein the assemblies are bound to a support.

24. The assemblies of Claim 23, wherein the support is selected from the group syphedex, silica gel, or 25    sepheros, polyacrylnitriles, filters, gold, silicon chips, silica gel.

25. The assemblies of Claim 1, wherein said color change is from blue or purple to pink, orange or yellow.

26. A test kit comprising a container incorporating the assemblies of Claim 1.

27. The test kit of Claim 26, wherein well structures contain the assemblies in suspension.
28. The test kit of Claim 26, wherein said kit container also incorporates instructions as to implementation of the test procedure.
29. A method of making the three dimensional polymeric multilayer assay assemblies of Claim 1, comprising,
- a) combining diyne monomers with ligands in an organic solvent,
  - b) evaporating the solvent,
  - c) adding an aqueous solution,
  - d) heating the solution above the main-phase transition temperature of the diyne monomers,
  - e) agitating the solution and cooling it to at least 4°C,
  - g) depositing the diyne-ligand in a polymerization chamber,
  - h) removing oxygen from the chamber,
  - i) polymerizing the diyne-ligand mixture short of the red phase.
30. The method of Claim 1, wherein in step a), the solvent is selected from chloroform, benzene, alcohol, cyclohexane, methylene chloride, acetonitrile, and carbontetrachloride.
31. The method of Claim 1, wherein the aqueous solution of step c) is selected from water, buffer solution, cell media, physiological saline, phosphate buffered saline, Trizma buffer, HEPES, and MOPS.
32. The method of Claim 1, wherein before the cooling in step e), the solution is filtered.

33. The method of Claim 1, wherein the diyne-ligand mixture is cooled at between 4° and -20° C for between 5 minutes and 5 hours
34. The method of Claim 33, wherein the mixture is cooled at between 0° and -15° C for between 5 and 20 minutes.
35. The method of Claim 34, wherein the mixture is cooled at between 0° and -5° C for between 5 and 12 minutes.
36. The method of Claim 1, wherein the diyne-ligand mixture is cooled during polymerization to between 1° and 22° C.
37. The method of Claim 36, wherein the diyne-ligand mixture is cooled during polymerization to between 16°-19° C.
38. The method of Claim 1, wherein step h) is accomplished by infusing said chamber with an inert gas.
39. The method of Claim 38, wherein said inert gas is argon or nitrogen.
40. The method of Claim 1, wherein the polymerization is achieved by U.V. irradiation using a pen ray lamp or a hand-held lamp.
41. The method of Claim 1, wherein the polymerization is achieved by gamma radiation, electron beam or X-rays, or other low energy ionizing source.
42. The method of Claim 1, wherein the polymerization is accomplished with an energy dose of 10-100 MJ/cm<sup>2</sup>.
43. The method of Claim 1, wherein the continues until the liposomes are in the blue or purple phase.



44. A method for the direct detection of an analyte in solution, comprising
- a) contacting the suspended assay assemblies of Claim 1 with a test sample, and
  - 5 b) observing the solution for a change in color to indicate the presence of an analyte.

## AMENDED CLAIMS

[received by the International Bureau on 28 June 1996 (28.06.96);  
original claims 30-33, 36, 38 and 40-43 amended;  
remaining claims unchanged (2 pages)]

27. The test kit of Claim 26, wherein well structures contain the assemblies in suspension.
28. ~~The test kit of Claim 26, wherein said kit container~~  
also incorporates instructions as to implementation of  
the test procedure.
29. A method of making the three dimensional polymeric multilayer assay assemblies of Claim 1, comprising,  
a) combining diyne monomers with ligands in an organic solvent,  
b) evaporating the solvent,  
c) adding an aqueous solution,  
d) heating the solution above the main-phase transition temperature of the diyne monomers,  
e) agitating the solution and cooling it to at least 4°C,  
g) depositing the diyne-ligand in a polymerization chamber,  
h) removing oxygen from the chamber,  
i) polymerizing the diyne-ligand mixture short of the red phase.
30. The method of Claim 29, wherein in step a), the solvent is selected from chloroform, benzene, alcohol, cyclohexane, methylene chloride, acetonitrile, and carbontetrachloride.
31. The method of Claim 29, wherein the aqueous solution of step c) is selected from water, buffer solution, cell media, physiological saline, phosphate buffered saline, Trizma buffer, HEPES, and MOPS.
32. The method of Claim 29, wherein before the cooling in step i), the solution is filtered.

33. The method of Claim 29, wherein the diyne-ligand mixture is cooled at between 4° and -20° C for between 5 minutes and 5 hours
34. The method of Claim 33, wherein the mixture is cooled at between 0° and -15° C for between 5 and 20 minutes.
35. The method of Claim 34, wherein the mixture is cooled at between 0° and -5° C for between 5 and 12 minutes.
36. The method of Claim 29, wherein the diyne-ligand mixture is cooled during polymerization to between 1° and 22° C.
37. The method of Claim 36, wherein the diyne-ligand mixture is cooled during polymerization to between 16°-19° C.
38. The method of Claim 29, wherein step h) is accomplished by infusing said chamber with an inert gas.
39. The method of Claim 38, wherein said inert gas is argon or nitrogen.
40. The method of Claim 29, wherein the polymerization is achieved by U.V. irradiation using a pen ray lamp or a hand-held lamp.
41. The method of Claim 29, wherein the polymerization is achieved by gamma radiation, electron beam or X-rays, or other low energy ionizing source.
42. The method of Claim 29, wherein the polymerization is accomplished with an energy dose of 10-100 MJ/cm<sup>2</sup>.
43. The method of Claim 29, wherein the continues until the liposomes are in the blue or purple phase.

AMENDED SHEET (ARTICLE 19)

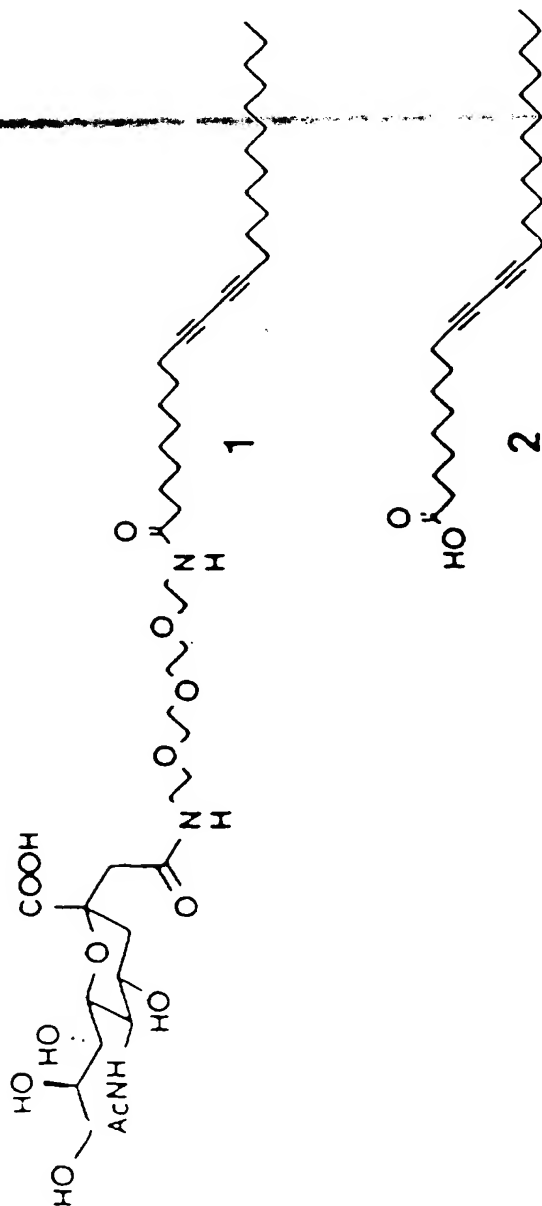


Fig. 1

2 / 2

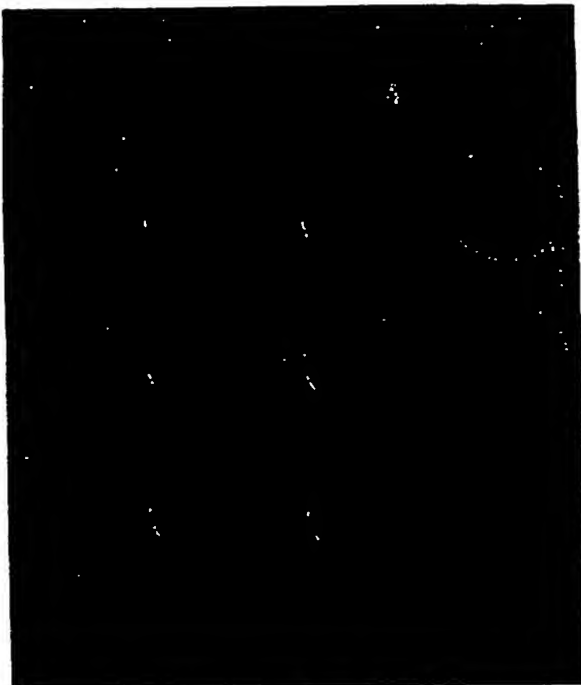


Fig. 2

## INTERNATIONAL SEARCH REPORT

International application No.  
PCT/US96/02017

**A. CLASSIFICATION OF SUBJECT MATTER**

IPC(6) : Please See Extra Sheet.

US CL : Please See Extra Sheet.

According to International Patent Classification (IPC) or to both national classification and IPC

**B. FIELDS SEARCHED**

Minimum documentation searched (classification system followed by classification symbols)

U.S. : 435/4, 5, 6, 7.1; 436/518, 528, 531; 422/55, 57; 427/2.11, 2.13, 2.23; 428/441, 442

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched  
None

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

APS, STN

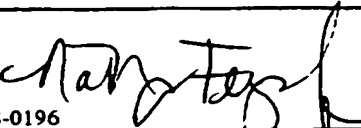
Search terms: polymer, polydiacetylene, biosensor, immunosensor, lipid, film, ethylmorpholin, pentacosadiynoic, analyte, color.

**C. DOCUMENTS CONSIDERED TO BE RELEVANT**

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X -- Y	CHARYCH et al. Specific Interaction of Influenza Virus with Organized Assemblies of Polydiacetylenes. Materials Research Society Symposium Proceedings. 1993, Volume 292, pages 153-161, see entire document.	1-5, 8-10, 12, 14-25, 29-43 ----- 6, 7, 11, 13, 26-28, 44
X -- Y	CHARYCH et al. Direct Colorimetric Detection of a Receptor-Ligand Interaction by a Polymerized Bilayer Assembly. Science. 30 July 1993, Volume 261, pages 585-587, see entire document.	1-5, 8-10, 12, 14-25, 29-43 ----- 6, 7, 11, 13, 26-28, 44

☒ Further documents are listed in the continuation of Box C. ☐ See patent family annex.

* Special categories of cited documents:	*T later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
*A* document defining the general state of the art which is not considered to be of particular relevance	*X* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
*E* earlier document published on or after the international filing date	*Y* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
*L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)	*G* document member of the same patent family
*O* document referring to an oral disclosure, use, exhibition or other means	
*P* document published prior to the international filing date but later than the priority date claimed	

Date of the actual completion of the international search 09 MAY 1996	Date of mailing of the international search report 23 MAY 1996
Name and mailing address of the ISA/US Commissioner of Patents and Trademarks Box PCT Washington, D.C. 20231 Facsimile No. (703) 305-3230	Authorized officer P. ACHUTAMURTHY  Telephone No. (703) 308-0196

Form PCT/ISA/210 (second sheet)(July 1992)\*

# INTERNATIONAL SEARCH REPORT

International application No.  
PCT/US96/02017

C (Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	PONS et al. The Optical Activity and Circular Dichroic Spectra of Diacetylenic Phospholipid Polymers. Biochimica et Biophysica Acta. 1982, Volume 693, pages 461-465, see entire document.	1-4, 15-22, 25, 27, 28-43
X	US 5,268,305 A (H. O. RIBI ET AL.) 07 December 1993, column 8, lines 6-7; column 9, line 56 to column 10, line 5; column 11, lines 6-8; column 16, line 49 to column 17, line 40.	1-5, 7, 15-44
Y		----- 6, 8-14
A	US 4,859,538 A (H. O. RIBI) 22 August 1989, see entire document.	1-44
A	YUAN et al. Colour Transition of Polyacetylene Multilayers Caused by Their Interaction with Several Kinds of Gases. Solid State Communications. 1991, Volume 80, Number 7, pages 493-495, see entire document.	1-44

# INTERNATIONAL SEARCH REPORT

International application No.  
PCT/US96/02017

**A. CLASSIFICATION OF SUBJECT MATTER:**  
**IPC (6):**

G01N 33/53, 33/543, 33/544, 33/545, 21/00, 31/22; B32B 17/10; B01J 13/00; B05D 3/00; C12Q 1/00, 1/68, 1/70

**A. CLASSIFICATION OF SUBJECT MATTER:**  
**US CL :**

435/4, 5, 6, 7.1; 436/518, 528, 531; 422/55, 57; 427/2.11, 2.13, 2.23; 428/441, 442